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# Molecular Formula Worksheet With Answers

**empirical and molecular formula worksheet** - microsoft word - empirical and molecular formula worksheets author: good, brian created date: 5/17/2013 11:16:32 am ... **chemistry - molecular formulas worksheet molecular formula ...** - chemistry - molecular formulas worksheet 1. the molecular formula of a substance is  $C_{12}H_{24}O_{10}$ . what is the empirical formula? 2. a compound has an empirical formula of  $n_2o_5$ . if the molecular formula mass is 432 g/mol, what is the molecular formula? 3. phosgene, a poisonous gas used in world war i, contains 12.1 % c, 16.2% o and 71.1 % ... **empirical and molecular formulas worksheet - empirical and molecular formulas worksheet**. objectives: • be able to calculate empirical and molecular formulas . empirical formula . 1) what is the empirical formula of a compound that contains 0.783g of carbon, 0.196g of hydrogen and 0.521g of oxygen? 2) what is empirical formula of a compound which consists of 89.14% au and 10.80% of o? **percent composition and molecular formula worksheet** - percent composition and molecular formula worksheet 1. what's the empirical formula of a molecule containing 65.5% carbon, 5.5% hydrogen, and 29.0% oxygen? 2. if the molar mass of the compound in problem 1 is 110 grams/mole, what's the molecular formula? 3. **empirical and molecular formulas worksheet** - what is molecular formula? 13) the compound methyl butanoate (an ester) smells like apples. its percent composition is: 58.8 % c, 9.9 % h, and 31.3 % o. if its molecular mass is 306 g/mole, what is the molecular formula? molarity, empirical and molecular formulas worksheet molarity **empirical and molecular formula worksheet** - empirical and molecular formula worksheet answer key. write the empirical formula for the following compounds. 1)  $C_6H_6$  2)  $CH_4$  3)  $C_8H_{18}$  4)  $C_4H_9$  5)  $WO_2$  6)  $WO_3$  7)  $C_2H_6O_2$  8)  $C_3H_8O$  9)  $X_3Y_{13}$  10)  $X_3$  **simplest formula worksheet - tracy unified school district** - molecular formula worksheet (empirical = simplest) molecular formula = 1. calculate empirical (simplest) formula (this may be given already). 2. find the molar mass of the empirical (simplest) formula. 3. divide molecular formula mass (given as molar mass or molecular formula mass) /empirical formula mass (molar mass). **chemistry computing formula mass worksheet** - find the formula mass of the following compounds. round atomic masses to the tenth of a decimal place. place your final answer in the formula mass column. chemistry computing formula mass worksheet problem set-up example: find the formula mass of  $Ca(NO_3)_2$  ca:  $1 \times 40.1 = 40.1$  n:  $2 \times 14.0 = 28.0$  o:  $6 \times 16.0 = 96.0$  \_\_\_ formula mass = 164.1 **empirical formula worksheet ii - wiggins school district** - empirical and molecular formula worksheet an oxide of chromium is found to have the following % composition: 68.4 % cr and 31.6 % o. determine this compound's empirical formula. the percent composition of a compound was found to be 63.5 % silver, 8.2 % nitrogen, and 28.3 % oxygen. determine the compound's empirical formula.

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